



## SYNTHESIS OF B-NOR-4-AZA-5 $\alpha$ -ANDROSTANE COMPOUND AS 5 $\alpha$ -REDUCTASE INHIBITOR

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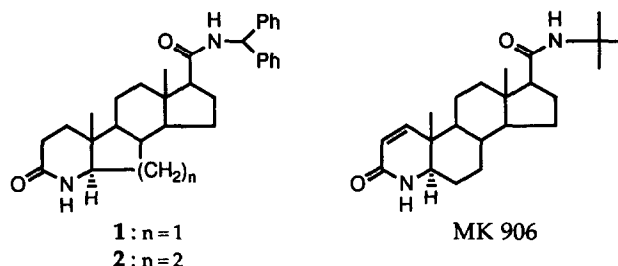
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**Abstract :** B-Nor-4-aza-5 $\alpha$ -androstane derivative **1** was synthesized. The compound **1** showed potent inhibition activity against testosterone 5 $\alpha$ -reductase.

Testosterone 5 $\alpha$ -reductase plays an important role in the pathology of benign prostatic hypertrophy. It converts testosterone in the prostate to 5 $\alpha$ -dihydrotestosterone, whose accumulation is known to cause hypertrophy of the prostate.<sup>1)</sup>

A series of 4-azasteroid compounds have already been synthesized, and they were found to have 5 $\alpha$ -reductase inhibition activity.<sup>2)</sup> Among them, N-(1,1-dimethylethyl)-3-oxo-4-aza-5 $\alpha$ -androst-1-ene-17 $\beta$ -carboxamide, MK-906, was recently launched as a drug for benign prostatic hypertrophy. These compounds have a nitrogen at the C-4 position and a backbone structure similar to that of testosterone. In relation to the structure-activity relationship we were interested in modifying the backbone structure, in particular the B-ring. Namely, we focussed attention on synthesizing a B-nor-4-aza-5 $\alpha$ -androstane derivative. The conformation of B-nor-4-aza-5 $\alpha$ -androstane, having a five-membered B ring, is not much different from that of 4-aza-5 $\alpha$ -androstane based on molecular model inspection. In addition, we have already disclosed that the 4-azaandrostane derivative **2**, having a 17-benzhydrylcarbamoyl moiety, has highly potent inhibition activity against testosterone 5 $\alpha$ -reductase.<sup>3)</sup> Therefore, the B-nor-4-aza-5 $\alpha$ -androstane derivative **1** with a benzhydrylcarbamoyl moiety at the C-17 position was selected as a target compound. In this report we describe a synthesis of **1** and its 5 $\alpha$ -reductase inhibition activity.

**Figure 1**



The compound **1** was synthesized from **3** by the route described below. First we selected secodicarboxylic acid **12**, having a B-nor structure, as a key intermediate. The B-nor structure was constructed according to a method similar to that described by Morisawa *et al.*<sup>4)</sup> Ester **3**<sup>2a)</sup> derived from Pregnenolone was acetylated to provide **4**, which was oxidized with ozone to give **5** (81% from **3**). Aldehyde **5** was cyclized to **6** with basic alumina, which was oxidized to yield carboxylic acid **7** (75% from **5**). Heating **7** with acetic anhydride and triacetin under reflux afforded **8** (66%), via a formation of  $\beta$ -lactone, followed by elimination of carbon dioxide. Deacetylation of **8** afforded ester **9**. Oxidation of **9** by the Oppenauer method, followed by hydrolysis gave enone **11** (80% from **8**). In order to introduce the 4-aza function, dicarboxylic acid **12** was synthesized. Oxidative cleavage of **11** with  $\text{NaIO}_4$  and  $\text{KMnO}_4$  in aqueous *tert*-butyl alcohol afforded **12** (28 %).

Next, the cyclization of dicarboxylic acid **12** to **13** was attempted. Upon heating **12** with ammonia, as is described for the synthesis of MK 906, the desired enamido compound **13** was not obtained. Therefore, we investigated preparing the saturated compound **14a** directly from **12**. Following the Leuckart method, heating **12** with formamide and formic acid, a cyclization occurred, to give B-nor-4-azaandrostane compound **14b** (39%). But the compound **14b** had undesired  $5\beta$ -H configuration. This was determined by the presence of Nuclear Overhauser Effect (NOE) between the hydrogen at the C-5 position and the methyl group at the C-19 position.

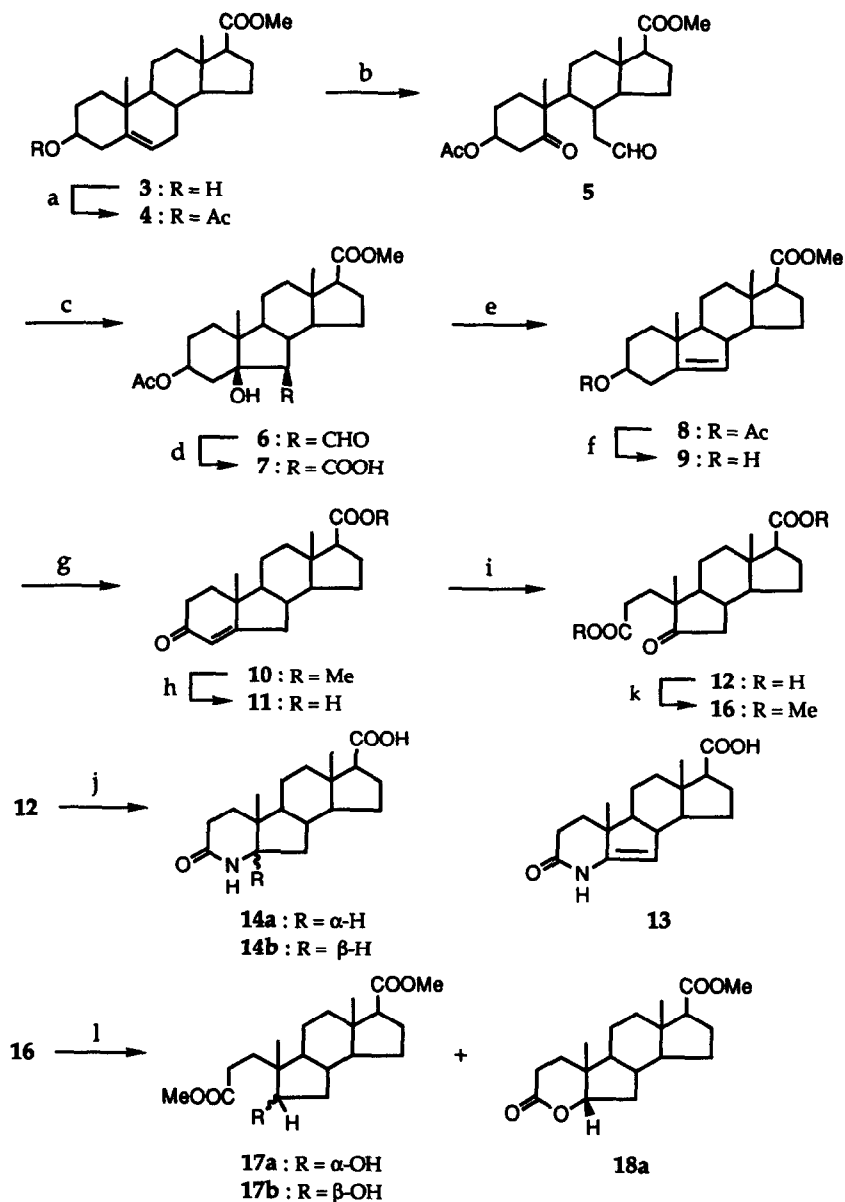
Then, after introducing a nitrogen function onto the B-ring of compound **16**, formation of the A-ring was accomplished. Dicarboxylic acid **12** was esterified to provide **16**. Diester **16** was reduced by  $\text{NaBH}_4$  in aqueous EtOH, to yield the desired  $\alpha$ -alcohol **17a** containing lactone **18a** together with  $\beta$ -alcohol **17b** (major product).<sup>5)</sup> A mixture of compounds **17a** and **18a** was converted, on treatment with *p*-TsOH, to a single product **18a**<sup>6)</sup> (4% from **16**). In order to prepare the  $5\beta$ -azide **26**, lactone **18a** was reduced with  $\text{LiAlH}_4$  to triol **19a**. Selective silylation of the primary hydroxy group in **19a** gave **20a**<sup>7)</sup> (53% from **18a**). The  $\alpha$ -alcohol **20a** was mesylated to afford **22**. Treatment of **22** with  $\text{NaN}_3$  afforded an azide **23** (19%) with the desired  $\beta$ -configuration at the C-5 position. Desilylation and oxidation of **23** gave dicarboxylic acid **25** (82%). Esterification of **25** gave diester **26** and reduction of **26** afforded directly a B-nor-4-aza- $5\alpha$ -androstane compound **27** (59% from **25**).

Finally, the 17-carbamoyl function was introduced. Treatment of **27** with  $\text{BCl}_3$  yielded carboxylic acid **28**.<sup>8)</sup> Amidation of **28** with DEPC and benzhydrylamine gave B-nor-4-aza- $5\alpha$ -androstane derivative **1** (30% from **27**).

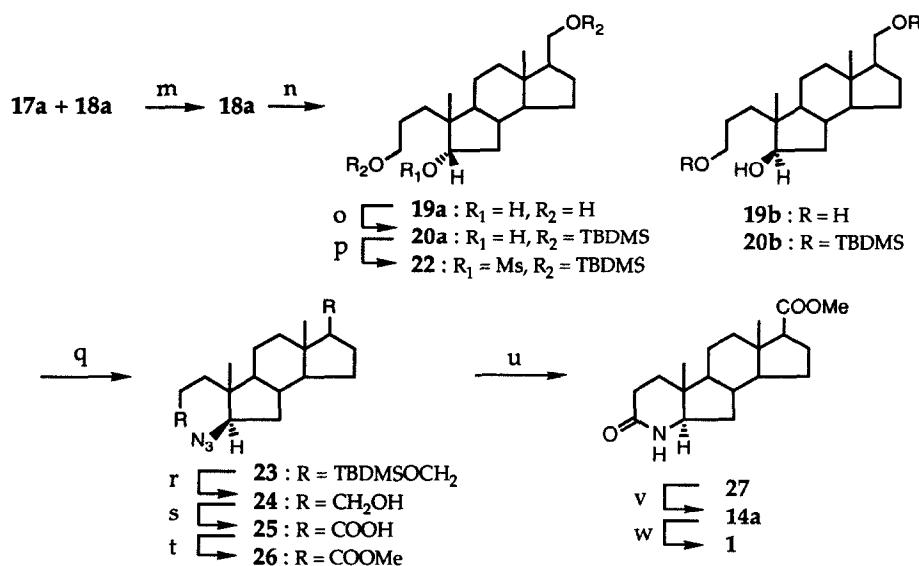
The *in vitro* inhibition of rat  $5\alpha$ -reductase was determined by using the standard method.<sup>9)</sup> The compound **1** had a 63% inhibition rate at  $10^{-8}\text{M}$  concentration. It was more potent than MK-906 (28% inhibition).

In conclusion, the compound **1**, which has a new B-nor-4-aza- $5\alpha$ -androstane structure, was synthesized, and it showed fairly potent inhibition activity against  $5\alpha$ -reductase.

Scheme 1



Reagents : (a)  $Ac_2O$ , pyridine, r.t. $\rightarrow$ 70°C ; (b)  $O_3$ ,  $CH_2Cl_2$ -MeOH, -78°C then Zn, AcOH ; (c) alumina (grade 3), benzene, r.t. ; (d) Jones reagent, acetone, 0°C ; (e)  $Ac_2O$ , triacetin, reflux ; (f)  $K_2CO_3$ , MeOH, 60°C ; (g)  $Al(i-PrO)_3$ , cyclohexanone, toluene, reflux ; (h) KOH, MeOH- $H_2O$ , reflux ; (i)  $KMnO_4$ ,  $NaIO_4$ ,  $Na_2CO_3$ ,  $t$ -BuOH- $H_2O$ , reflux ; (j) formamide, formic acid, 180°C ; (k)  $CH_2N_2$ , EtOAc, r.t. ; (l)  $NaBH_4$ , EtOH- $H_2O$ , r.t.



Reagents : (m) *p*-toluenesulfonic acid, benzene, reflux ; (n)  $LiAlH_4$ ,  $Et_2O$ -THF,  $0^\circ C \rightarrow r.t.$  ; (o) *t*-BuMe<sub>2</sub>SiCl, imidazole, DMF, r.t. ; (p) MsCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, r.t. ; (q) NaN<sub>3</sub>, DMF,  $110^\circ C$  ; (r) AcOH, THF-H<sub>2</sub>O, r.t.  $\rightarrow 50^\circ C$  ; (s) Jones reagent, acetone, r.t. ; (t) CH<sub>2</sub>N<sub>2</sub>, CHCl<sub>3</sub>-MeOH, r.t. ; (u) H<sub>2</sub>, PtO<sub>2</sub>, Molecular Sieves 3A, MeOH, r.t. ; (v) BCl<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, r.t. ; (w) benzhydrylamine, DEPC, Et<sub>3</sub>N, r.t.

## References and Notes

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- Yield of **17b** was 43%. The formation ratio of **17a**, **17b**, and **18a** = 20, 9, and 71 based on <sup>1</sup>H-NMR.
- Stereochemistry at the C-5 position in **18a** was assigned based on the NOE experiment. The NOE was observed between 5β-H and 19-methyl group in **18a**.
- The undesired isomer **17b** was also converted to the desired **20a** : By using a similar sequence of reactions to that described for the synthesis of **20a** from **18a**, **17b** gave **20b**. Oxidation and then reduction of **20b** gave **20a** (35%) along with **20b** (64%).
- The lactam ring of **27** was labile to the general hydrolysis condition used for the hydrolysis of **17**-ester.
- The biological test method is described in reference 3.

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